

Energy Auditing and Waste Heat Recovery for Dry Type Cement Kiln System in Ethiopia: A Case Study

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Abstract: The aim of this study is energy audit and heat recovery of a dry type technology of cement industry in Ethiopia. The kiln has a capacity to produce 2000 ton-clinker per day. By considering 12 months data for mass and energy balance, auditing energy is performed. From the result about 35% of the total input energy was being lost through hot flue gas (19.53%) and cooler stack (16.22%). By using waste heat recovery steam generator (WHRSG) about 5 MW (6.13%) of the total input energy could be recovered. The whole electrical energy usage of the kiln system safely used from the generated power. So, the factory could keep 800,000 USD/Year by using the generated power for production of clinker.

Keywords: Cement plant; Kiln system; Energy audit; Waste heat recovery

INTRODUCTION

Cement production is an energy intensive process, consuming about 4 GJ per ton of cement product. In order to produce one ton of clinker a minimum of 1.6 GJ heat is required [Liu et al., 1995]. However, now days, it is about averagely 2.95 GJ energy is consumed per ton of cement for advanced kilns, but in some countries, the consumption exceeds 5 GJ/ton. For example, the average energy requirement of Chinese key plants to produce clinker is 5.4 GJ/ton [Khurana et al., 2002]. The energy audit is the most effective procedures for good and well energy management program [Pahuja, 1996]. The main goal of energy audits is to give a correct account of energy requirement and use analysis of different components

and to provide the detailed information needed for determining the possible opportunities for energy conservation. There are technical ways to improve overall kiln efficiency like waste heat recovery from hot gases and hot kiln surfaces [Pahuja, 1996 and Kamal, 1997]. However, it is difficult to express a detailed thermal analysis of rotary kiln systems in the cement plant.

The main objective of this study is about the energy audit and ways of heat recovery mechanism of the kiln system, by using WHRSG and also giving directions on how to increase the efficiency of the system and reduction of energy consumption on Messebo Cement Plant line-1 in Ethiopia. First energy auditing is performed and then, application of WHRSG for heat recovery mechanisms are discussed.



Figure 1. Methods for waste heat recovery mechanism

PLANT AND PROCESS DESCRIPTION AND DATA GATHERING

The cement plant in this paper is one of the major cement producing industry in Ethiopia. The plant is installed at an altitude of about 2200 meters above sea level [Bayray et al., 2013]. The plant produces

2,000 tons/day and uses dry type technology with five stage cyclone preheaters and pre-calciner kilns and grate cooler. The rotary kiln has 3.8m diameter and 57 m length. Specifically energy consumption of the plant is 3.7 GJ per ton of clinker and the average coal consumption of the kiln system for 12 months is 230 tons/day. The system of the cement production mainly includes following steps:

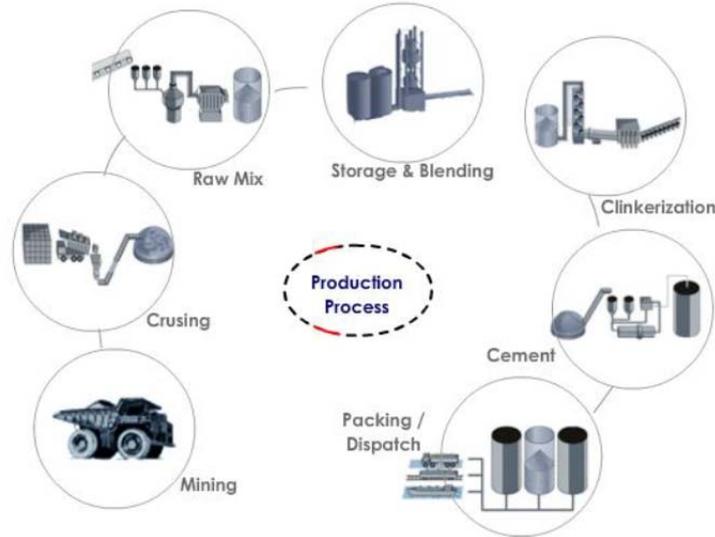


Figure 2. The production process of cement.

From Messebo Cement Plant line 1 large number of data have been collected for a long time and different measurements are used in 12 months and averaged and interpolated values are used in this study.

ENERGY AUDITING AND HEAT RECOVERY

3.1 Assumptions

The following assumptions are made to analyze for auditing energy.

- a) The system is steady state and steady flow.
- b) Ambient temperature are constant throughout the study i.e. $T_0 = 297K$.
- c) The composition of raw material and coal material and feed rate of both are constant.
- e) The velocity of atmospheric air is $< 3 \text{ m/s}$.
- f) The temperature of kiln shell is constant throughout the study.
- g) Consider the gases inside the kiln as an ideal gas.

3.2 Mass Balance

It is usual to define mass/energy data per kg clinker production per unit time.

Table1. Raw materials and clinker mixtures and their amount in percent

Mixtures	Raw Materials (%)(kg/kg)	Clinkers (%)(kg/kg)
SiO ₂	13.4	21.1
Al ₂ O ₃	3.2	4.08
Fe ₂ O ₃	2.4	4.01
CaO	42.35	66.44
MgO	1.71	2.42
SO ₃	0.4	1.15
K ₂ O	0.27	0.5
Na ₂ O	0.09	0.3
H ₂ O	0.01	---
Organics	0.8	---
Ignition loss	35.4	---
Overall	100%	100%

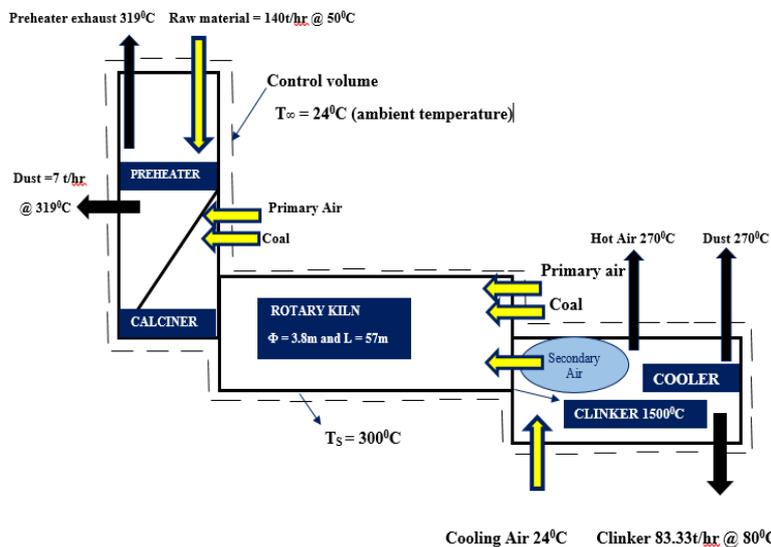


Figure 3. Control volume, various streams and components for kiln system.

Energy Balance

Based on the collected data, the energy balance is applied to the kiln system. By using Zur-Strassen equation the energy used for clinker formation can be found [Peray, 1979 and Tahsin, 2004].

$$\text{Clinker formation energy} = 17.196(\text{Al}_2\text{O}_3) + 27.112(\text{MgO}) + 32(\text{CaO}) - 21.405(\text{SiO}_2) - 2.468(\text{Fe}_2\text{O}_3)$$

By using Dulong's formula the GCV (Gross Calorific Value) of the coal is found [R.K. Patil et al., 2013].

$$\text{GCV} = 337 \times C + 1442 (H - O/8) + 93 \times S$$

Table2. Percentage composition of Coal

Element Content	Percentage (%)
C	73
H	3.5
O	6
N	1.75
S	1.59
Ash	3.01
Moisture	0.309
Volatile	4.67
Fixed Carbon	9.05

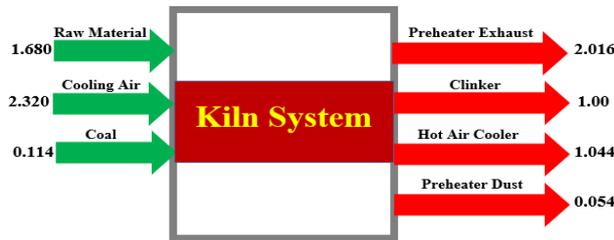


Figure 4. Mass balance of the kiln system.

Heat Inputs

1. Combustion of Coal (95.22%)

$$Q_1 = M_c H_c = 3358.25 \text{ kJ/kg-clinker}$$

$$M_c = 0.114 \text{ kg/kg-clinker}$$

$$H_c = 29,458.32 \text{ kJ/kg}$$

2. Sensible heat by coal (0.19 %)

$$Q_2 = M_c H_c = 6.9 \text{ KJ /kg-clinker, } H_c =$$

$$CT \quad C = 1.15 \text{ kJ/kg}^0\text{C} \quad T = 50^0\text{C}$$

3. Heat by Raw material (2.05%)

$$Q_3 = M_{Rm} H_{Rm} = 72.24 \text{ kJ /kg-clinker,}$$

$$H_{Rm} = CT \quad C = 0.86 \text{ kJ/kg}^0\text{C} \quad T = 50^0\text{C}$$

4. Organics in the kiln feed (0.54%)

$$Q_4 = B S h_{os} = 18.93 \text{ kJ /kg-clinker,}$$

$$B = 0.10, S = 0.009, h_{os} = 21,036 \text{ kJ/kg}$$

5. Heat due to cooling air (2%)

$$Q_5 = M_{ca} H_{ca} = 70.5 \text{ kJ /kg-clinker, } H_{ca} =$$

30 kJ/kg

6. Total heat inputs

$$Q_T = \sum_{i=1}^5 Q_i = 3526.82 \text{ kJ/kg (100\%)}$$

Heat Outputs

1. Kiln exhaust gas (19.53%)

$$Q_6 = M_{ex} C_{ex} T_{ex} = 688.76 \text{ kJ/kg-clinker}$$

$$C_{ex} = 1.071 \text{ kJ/kg}^0\text{C, } T = 319^0\text{C}$$

2. Heat loss by dust (0.42%)

$$Q_7 = M_{DhD} = 14.85 \text{ kJ/kg-clinker} \quad h_D = 275 \text{ kJ/kg}$$

3. Hot air from cooler (16.22%)

$$Q_8 = M_{Ha} h_{Ha} = 571.87 \text{ kJ/kg-clinker}$$

$$h_{Ha} = 547.77 \text{ kJ/kg (@ } T = 270^0\text{C)}$$

4. Clinker discharge (2.02%)

$$Q_9 = M_{clinker} h_{clinker@80^0\text{C}} = 71.09 \text{ kJ/kg-clinker}$$

5. Clinker formation (51.04%)

$$Q_{10} = 1800 \text{ kJ/kg-clinker}$$

6. The combined effects of convection and radiation on the kiln shell is calculated below and has the heat lost from kiln surface (5.27%)

Kiln surface radiation

$$Q = \sigma \epsilon A (T_s^4 - T_{\infty}^4) / 1000 * M_{clinker} = 130 \text{ kJ/kg-clinker; } T_s = 3000\text{C, } T_{\infty} = 2400\text{C}$$

$$M_{clinker} = 83333.33 \text{ kg/3600sec} = 23.15 \text{ kg/sec;}$$

$$A_{kiln} = \pi D L = \pi * 3.8 * 57 = 680.5 \text{ m}^2$$

Kiln surface convection

$$Q = h A (T_s - T_{\infty}) / 1000 * M_{clinker} = 55.93 \text{ kJ/kg-clinker}$$

$$h_{con} = K_{air} * Nu / D_{kiln}$$

$$Re = 375,617.8 \text{ (} V_{air} = 3 \text{ m/s), } Nu = 738 \text{ } T_f = 162^0\text{C (film temp.) [Cengel, 2003]}$$

$$\text{Hence; } Q_{11} = 4.3 \text{ MW} = 185.94 \text{ kJ/kg-clinker}$$

7. Heat exhaust from pre-heater surface due to radiation (0.23%)

$$Q_{12} = \sigma \epsilon A_{ph} (T_s^4 - T_{\infty}^4) / 1000 M_{clinker} = 8.11 \text{ kJ/kg-clinker}$$

$$A_{ph} = 264 \text{ m}^2, T_{phs} = 393 \text{ K, } T_{\infty} = 297 \text{ K,}$$

8. Heat exhaust from pre-heater surface due to natural convection (0.3%)

$$Q_{13} = h_{ncon} A_{ph} (T_s - T_{\infty}) / 1000 M_{clinker} = 10.61 \text{ kJ/kg-clinker}$$

$$h_{ncon} = K_{air} * Nu / L_{ph}; Ra = 5.35 * 10^{13}, Nu = 0.1 (Ra)^{1/3} Nu = 3768.06 \text{ (Ref. [10]), } T_f = 72^0\text{C (film temp.)}$$

9. Heat exhaust from cooler surface due to radiation (0.11%)

$$Q_{14} = \sigma \epsilon A_c (T_s^4 - T_{\infty}^4)$$

$$/ 1000 M_{clinker} = 3.8 \text{ kJ/kg-clinker}$$

$$A_c = 144 \text{ m}^2, T_{Cs} = 383 \text{ K, } T_{\infty} = 297 \text{ K, } M_{clinker} = 23.15 \text{ kg/s}$$

10. Natural convection from cooler surface (0.07%)

$$Q_{15} = h_{ncon} A_c (T_s - T_{\infty}) / 1000 M_{clinker} = 2.6 \text{ kJ/kg-clinker}$$

$$h_{ncon} = K_{air} * Nu / L_C, Ra = 7.89 * 10^{12}, Nu = 0.1(Ra)^{1/3} Nu = 1991, L_C = 12, T_f = 67^{\circ}C$$

(film temp.)

11. Moisture in raw material and coal (0.69%)

$$Q_{16} = m_{water} (h_{fg@50^{\circ}C} + h_{g@319^{\circ}C} - h_{g@50^{\circ}C}) = 24.41 \text{ kJ/kg-clinker}$$

$$h_{fg@50^{\circ}C} = 2384 \text{ kJ/kg}, h_{g@50^{\circ}C} = 2591 \text{ kJ/kg}, h_{g@319^{\circ}C} = 2700 \text{ kJ/kg}, m_{water} = 0.0098 \text{ kg/kg-clinker}$$

12. Unaccounted heat losses (4.1%)

$$Q_{17} = 144.78 \text{ kJ/kg-clinker}$$

13. Total heat output

$$Q_T = \sum_{i=1}^{17} Q_i = 3526.82 \text{ kJ/kg-clinker (100\%)}$$

HEAT RECOVERY FROM THE KILN SYSTEM

The overall system efficiency is given by $\eta = (\text{clinker formation}) / (\text{total heat input}) = 1800 / 3526.82 = 0.5104$ or 51.04% which is consider as low since some kiln systems use the current technology that have 55% efficiency. In order to increase the performance of the kiln system some of the heat losses should be recovered. The recovered heat energy is used for various services, like boiling of hot water as well as electricity generation. There are two major heat loss sources that should be recovered by using WHRSG. These are preheater exhaust gas (19.53%) and hot air from cooler vent (16.22%).

Design of Waste Heat Recovery Steam Generator (WHRSG)

Based on these two highest waste heat sources, our waste heat recovery system is designed. The main tapping spots are the exhaust pipe before it enters into the induced draught fan and after the first cyclone from the top for the kiln exhaust and the pipe just after the hot air leaves the cooler stack and before it goes the heat exchanger and bag filter for the cooler side. The exhaust gas from the preheater is used to

dry the raw material and coal before milling. Therefore, the acid dew point temperature of preheater gases, should not be less than 1600C (The acid dew point is the temperature at which the vapor condenses and for coal range it is about 1600C) [Weston, 1992].

The exhaust gas from the preheater is 3190C and the temperature of the air exhausted from the cooler stack is 2700C. These two streams would be going through a waste heat recovery steam generator (WHRSG), and the available heat energy is transferred to water via the WHRSG as shown in Fig. 3. The available waste energy could generate steam. This steam is used to run a steam turbine to drive electrical generator. The generated electricity could cover the electrical consumption of the kiln system.

The available energy from the gas streams have to be determined, to determine the size of the generator by calculating the steaming rate and pressure. Hence, it is given by:

$$Q_{WHRSG} = \eta Q_{Available}$$

The available energy that can be harnessed from the preheater exhaust gas and cooler vent hot air is given by:

$$\dot{Q}_{available} = \dot{Q}_{ph\ ex} + \dot{Q}_{air}$$

$$\dot{Q}_{available} = [m_{ph\ ex} (h_{ph\ ex1} - h_{ph\ ex2}) + m_{air} (h_{air1} - h_{air2})] \dot{m}_{clinker} = 12,720 \text{ KW}$$

Where $h_{air1@T=270^{\circ}C} = 277 \text{ kJ/kg}$, $h_{air2@T=80^{\circ}C} = 79.53 \text{ kJ/kg}$, $h_{ph\ ex1@319^{\circ}C} = 341.65 \text{ kJ/kg}$ and $h_{ph\ ex2@160^{\circ}C} = 171.36 \text{ kJ/kg}$ [NSRDS, 1971].

Consider the overall efficiency of the system is 90% of the total available heat. Hence, $Q_{available} = 11,450 \text{ kW}$. This high amount of energy can be recovered but it depends on technological advancement of the system or other physical, technical and operational factors.

In order to utilize this energy, we have to select the size of the steam turbine generator. A steam turbine is a rotating machine used to produce net amount of work done. Hence, proper maintenance and clean dry steam is needed, for good operation.

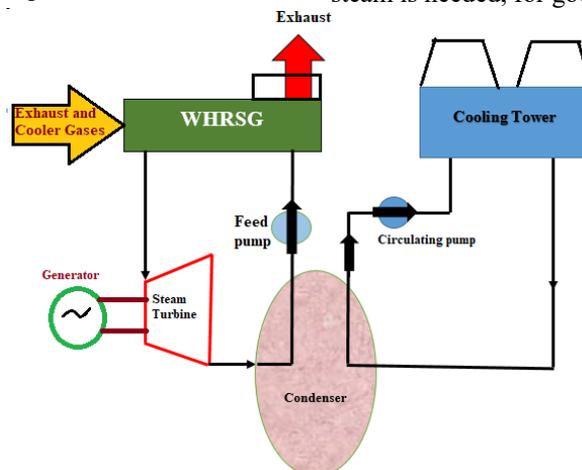


Figure 5. The use waste heat recovery steam generator (WHRSG) process diagram.

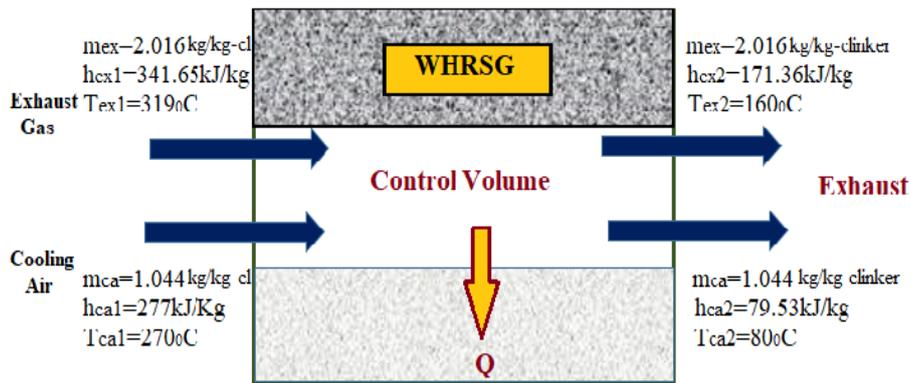


Figure 6. The process diagram of the waste heat recovery steam generator (WHRS).

Consider a turbine and condenser pressure is 8 bars and 10 kPa respectively, so that the net power obtained from the turbine, is almost 5000 kW. By considering the useful power generated is 5000 kW, the expected savings will be based on the load reduction of 5000 kW. Considering 8000 h of usage, the expected savings become;

$$\text{Energy saved} = (\text{Power generated}) * (\text{hours of usage}) \\ = 5000\text{kW} * 8000\text{hr/yr.} = 4 * 10^7 \text{kWh/yr.}$$

The energy cost of the current factory under the analysis is 0.0211USD/kWh [Berhe, 2012] and hence, the expected cost savings is given by:

$$\text{Cost savings} = 0.0211\text{USD/kWh} * 4 * 10^7 \text{kWh/yr.} = 844,000\text{USD/yr.}$$

By considering the average labor and maintenance costs is 40,000 USD annually, the saving becomes 800,000 USD/year.

The implementation cost of this additional system could be the purchase price of important equipment and its installation. The power generation unit may require additional cost for maintenance. By considering the whole system as shown in Fig. 5, a budget is estimated between 1,500,000 – 2,000,000USD, including shipping and installation. Hence, by using a rough estimation a simple payback period becomes:

$$\text{Simple payback period} = \frac{\text{implementation cost}}{\text{annual cost savings}} \\ = \frac{2,000,000}{800,000} = 2.5\text{yr} = 30 \text{ months}$$

The energy savings by using a WHRS system is also used to increase the performance of the overall system. Generally, it should be noted that these analysis shows a rough estimation and it depends on plant conditions and other economic factors.

Use of Waste Heat to Pre-Heat the Raw Material

Preheating raw material and coal before the clinkering process is the most effective method of recovering waste heat in cement plants. In this

method gas streams are directed into the raw material just before the grinding mill. This would result increasing the efficiency of grinding the raw material in addition to increasing of its temperature. However, in most cement plants, the fresh raw material and coal taken from the mill is not directly enter to the kiln, and therefore, the temperature increase of the raw material is not useful because it will be stored in silos for some time just before entering the clinkering process. On the other hand, some plants may have only kiln systems instead of grinding systems. In this case some additional modifications have to be made in the plant [Tahsin, 2004 and 2011].

In this study our cement plant has grinding mill and applying preheating technique could save high amount of energy. The main purpose of pre-heating raw material in the mill is drying the material, because it is moisty in nature. By considering, the moisture content of the raw material i.e. 6% and a mass flow rate of water of 2.315kg/s (0.1 kg/kg-clinker) enter into the mill. Mixing of the two main hot gas streams would result in a single gas flow at about 3000C, as shown in Fig. 7. By applying the conservation of mass and energy principle for the mill (neglect heat losses) which leads as to increase of temperature of the raw material by 860C, but the gas stream cools down to 1700C. It is made evident that the majority of the useful energy have to be used to heat the water from 24 to 1100C, and to vaporize completely it at this temperature [13, 14]. The energy balance for the mill system is given by:

$$Q_{\text{gas flow in}} + Q_{\text{moist raw material}} = Q_{\text{water}} + Q_{\text{gas, out}} + Q_{\text{dry raw material}}$$

$$Q_{\text{gas, out}} = 23.15[(3.06 * 321.3) + (1.78 * 20)] - 2.315(461.3 - 100.7 + 2230.2) - 23.15(1.68 * 94.6)$$

$$Q_{\text{gas, out}} = 13,908\text{kW} \quad h_{\text{gas, out}} \approx 175\text{kJ/kg} \quad T_{\text{gas, out}} \approx 170^\circ\text{C}$$

$$\text{Where, } h_f@24^\circ\text{C} = 100.7\text{kJ/kg, } h_f@110^\circ\text{C} = 461.30\text{kJ/kg, } h_{fg}@110^\circ\text{C} = 2230.2\text{kJ/kg.}$$

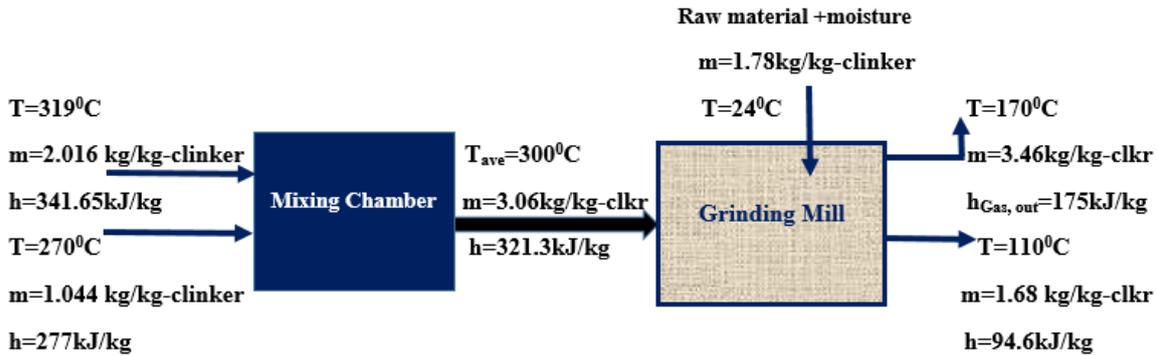


Figure 7. Mass and energy balance for grinding mill.

CONCLUSION

The main aim of this study is to determine the heat lost and discussing the recovering mechanisms. By using the actual plant operational data mass and energy analysis were carried out. From the results which we are obtained, the system efficiency is 51% and the major heat loss sources are preheater exhaust (19.53%) and cooler stack (16.22%) of the total input energy. In order to recover these heat lost WHRSG system is selected and designed. By applying this recovering mechanism 5 MW of energy could be recovered, which means 6.12% of the total input energy is recovered. The anticipated payback period for the systems is to be less than 2.5 yr. Electricity Production with own power using heat lost from the process will enable as to reduce electrical consumption from national grid supply and also decreases clinker production cost. Hence, in order to minimize energy consumption and manufacturing costs WHRSG system have to be implemented in the design of new plants. It is recommended that Ethiopian government should have to give great concern for cement industries by supporting this WHRSG implementation through exemption of tax and other incentives.

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NOMENCLATURE

M	Mass
Q	Heat
C	Specific heat capacity
σ	Stefane Boltzman constant = $5.67 \cdot 10^{-8}$
$W/m^2 K^4$	
ϵ	Emissivity of the surface (for oxidized surface = 0.78)
T_{∞}	Ambient temperature
Nu	Nusselt number.

Ra	Rayleighs number
h_{ncon}	Natural convective coefficient
h_{fg}	Enthalpy of moisture per unit mass(kJ/kg)
h_g	Enthalpy of saturated vapor per unit mass(kJ/kg)
k	Thermal conductivity of the material
A	Area of the rotary kiln
E	Radiative heat transfer (heat flux)
h	Heat transfer coefficient
F_{12}	View factor that takes (=1)
η	The overall system efficiency

REFERENCES

- Bayray M, Mustefa A, Yohannes F, Kiros H, Haileslasie A, Gebray P, et al. Wind energy data analysis and resource mapping of geba catchment, north Ethiopia. *Wind Engineering*. 2013; 37:333-46.
- Berhe H. Feasibility study for generation of electric power from WHRPG of the two kiln exhaust gases of Messebo cement factory. Mekelle: Mekelle University; 2012.
- Cengel YA. *Heat transfer—A practical approach*. 2nd ed. New York, NY: McGraw-Hill; 2003.
- Kamal K. Energy efficiency improvement in the cement industry. Seminar on Energy Efficiency, organized by ASSOCHAM-India and RMA-USA, January 1997.
- Khurana S, Banerjee R, Gaitonde U. Energy balance and cogeneration for cement plant. *Appl Therm Eng* 2002; 22:485-94.
- Liu F, Ross M, Wang S. Energy efficiency of China's cement industry. *Energy* 1995; 20(7):669-81.
- NSRDS Thermodynamic tables Based on data from the JANAF Tables, NSRDS-NBS-37, 1971, and revisions published in *Journal of Physical and Chemical Reference Data* through 1982.
- Pahuja A. Energy auditing and monitoring in the cement industry. *Energy Conservation and Environmental Control in Cement Industry*, vol. 2, Part 2, Published by Akademia Books International, ISBN: 81-85522-05-7, 1996. p. 670- 94.
- Peray KE. *Cement manufacturer's handbook*. New York, NY: Chemical Publishing Co., Inc.; 1979.
- R. K. Patil, M. P. Khond, L. G. Navale, "Thermal Energy Assessment of Indian Cement Plant Specially Relates To

- Rotary Kiln”, Journal of Mechanical and Civil Engineering (JMCE),2013; pp 23- 31.
- Tahsin Engin, Energetic and exergetic assessments of a cement rotary kiln system Vedat Ari, 10 March, 2011
- Tahsin Engin, Energy auditing and recovery for dry type cement rotary kiln systems–A case study Vedat Ari, 2July 2004
- Weston KC. Energy conversion: West Publishing; 1992.